Synthesis and Structure Elucidation of the Condensation Products Between Thiophene Dicarbaldehydes and Aromatic Amines. Potential Analgesic and Anti-inflammatory Agents

F. Benachenhou and M. A. Mesli

Département de Chimie, Centre Universitaire de Sidi-bel-Abbes, Algérie

M. El Borai

Faculty of Science, University of Tanta, Tanta, Egypt

B. Hanquet and R. Guilard*

Laboratoire de Synthèse et d'Electrosynthèse Organométallique, Associé au C.N.R.S. (U.A.33), Faculté des Sciences "Gabriel", 6, Bvd Gabriel 21100 Dijon, France Received July 10, 1987

Thiophene-3,4-dicarbaldehyde 1 reacts in the presence of 2-mercaptoethanol to yield N-aryl-5,6-dihydro-4-oxo-4H-thieno[3,4-c]pyrroles 2 and N-aryl-4-arylimino-5,6-dihydro-4H-thieno[3,4-c]pyrroles 3, while thiophene 2,3-dicarbaldehyde 4 reacts with aromatic amines to give N-aryl-5,6-dihydro-6-oxo-4H-thieno[2,3-c]pyrroles 5 in good yields. Labeling experiments and nmr spectral analysis give evidences for the possible reaction mechanism.

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In the last decade, research on analgesic and antiinflammatory agents containing an isoindoline moiety, A, has led to various types of aryl-substituted compounds [1-3], but to our knowledge no five-membered-ring heterocycle analogue, B, has been reported.

Part of this is probably due to the commercial availability of phthalaldehyde. Heterocyclic ortho-dicarbaldehydes usually require multiple-step synthesis [4-5] or low yields are obtained [6]. The reactions involving phthalaldehyde and amines are pretty easy and gave good yields under dilute conditions [7]. However some side-reactions lead to other condensation products and/or to polymeric materials [8], depending on molar ratios of the reactants and their concentrations.

In this paper we will report the syntheses, characterization, and mechanism of formation of N-aryl-5,6-dihydro-4-oxo-4H-thieno[3,4-c]pyrroles 2, N-aryl-4-arylimino-5,6-dihydro-4H-thieno[3,4-c]pyrroles 3, and N-aryl-4,5-dihydro-6-oxo-4H-thieno[2,3-c]pyrroles 5.

Thiophene-3,4-dicarbaldehyde 1 does not react with aromatic amines under the conditions in which phthalaldehyde gave isoindolines. Even with increasing the temperature, 1 remains unreacted. The compounds 2 were

produced in high yields using a four-fold molar excess of 2-mercaptoethanol.

The di-adducts 3 were obtained with an excess of amine (see Scheme 1). All reactions were carried out in dilute solutions in order to avoid the formation of polymeric material (see Experimental).

Scheme 1

Compound 1 reacted with aromatic amines to give 2 under the conditions described above. But when 1,2-ethanedithiol was used, the carbonyl protected compound 6 was isolated in the case of p-toluidine. This result demonstrates that the chelating agent favors the formation of 2 (see Scheme 2).

The N-aryl-5,6-dihydro-4-oxo-4H-thieno[3,4-c]pyrroles 2 have been obtained in high yields (69 to 79%) from various aromatic amines. Compounds 3 were produced in lower yields (40 to 66%).

Either in the presence or absence of 2-mercaptoethanol thiophene-2,3-dicarbaldehyde 4 reacted with aromatic amines in refluxing xylene leading to the N-aryl-5,6-dihy-

Scheme 2

dro-6-oxo-4*H*-thieno[2,3-c]pyrroles **5** (see Scheme 3). Even with an excess of amine the formation of the di-adduct was not observed.

Scheme 3

The structure elucidation of compounds 2 and 5 was easily achieved on the basis of their ir and nmr data. The carbonyl absorptions arose in the range 1670-1690 cm⁻¹ and the imine stretching bands were found in the 1638-1645 cm⁻¹ region.

The nmr signals for all the thiophene protons appeared as two sharp resonances between 6.89 and 7.84 ppm, with the expected coupling constants, and the cyclic methylene protons appeared between 3.86 and 4.77 ppm (see Experimental).

For the unsymmetrical series, two isomers C and D, are theoretically obtainable but only one of them was isolated from the reaction mixture.

Firstly, the pmr spectroscopic data were not sufficiently explicit to determine whether the structure of the product is **C** or **D**. Secondly, it was not possible for the *N*-benzyl derivatives of **5** to uniquely assign the structure on the basis of chemical shift expectations for the two methylene proton resonances appearing at 4.1 and 4.75 ppm.

In order to solve the isomer structure problem and to give the nmr signals assignment, deuterium labelled compounds were prepared. Moreover, the analysis of the labelled products gave significant data about the mechanism of their formation.

Scheme 4 displays the profile of the pmr spectra in the 4-5 ppm region.

Scheme 4 CHO SCHO CHO SCHO SCHO

From the spectra (a) and (b) it is obvious that the methylene in the five-membered-ring is the high-field singlet. Analysis of spectrum (c) proves that the deuterium atom was lost during the reaction because the two singlets were still present. The definitive proof is afforded by examination of the fourth spectrum (d). The deuterium atom in the 3 position of the dicarbaldehyde was retained through the transformation and the resonance at 4.18 ppm looked like a broad triplet due to the coupling with the deuterium atom. Therefore, the structure of all the compounds and the mechanism of their formation may be deduced from these data. Scheme 5 summarizes the proposed mechanism.

The proof that the reaction began by the condensation with the aldehyde group in position 3 was given by the retention of the deuterium atom in that position. Otherwise, it would be lost during the 1,3-hydride shift [9].

EXPERIMENTAL

All melting points are uncorrected and were obtained with a Kosler hot-stage apparatus. The ir spectra were recorded on a Perkin-Elmer 580B instrument on 1% potassium bromide disks. The nmr spectra (δ scale) were recorded at the Burgundy's University nmr facility (CEREMA) on a Bruker WM400 spectrometer (400 MHz, deuteriochloroform, TMS as internal standard, J in Hz). Multiplicities are given as follows: singlet (s), doublet (d), triplet (t), quartet (q), and multiplet (M). Analyses were performed by the "Service Central du C.N.R.S.".

General Procedure for the Preparation of 2, 3, and 5.

N-Aryl-5,6-dihydro-4-oxo-4H-thieno[3,4-c]pyrroles 2.

A solution of the aromatic amine (0.00275 mole) in 20 ml of anhydrous xylene was added dropwise under reflux to a stirred solution of 1 (0.0025 mole, 0.35 g) and (0.01 mole, 0.7 ml) of 2-mercaptoethanol in 200 ml of anhydrous xylene. After 4 hours under reflux the reaction mixture was quenched with water, and the organic layer was washed with water and dried over magnesium sulfate. The solvent was removed under vacuum and the residue chromatographed on a silica gel column. Recrystallization afforded yellow-to-red crystals.

N-Phenyl-5,6-dihydro-4-oxo-4H-thieno[3,4-c]pyrrole 2a.

Compound 2a was obtained using the general procedure described above. Elution with ether/hexane (1/1) mixture and recrystallization from methylene chloride/hexane give 2a in 69% yield, mp 148-149°; ir: ν C=0 1691.5 cm⁻¹; pmr: δ 7.84 (d, H₃, 2.3 Hz), 7.18 (dt, H₁, 2.3 and 1.1 Hz), 7.20-7.77 (M, C₆H₂), 4.77 (d, CH₂, 1.1 Hz).

Anal. Calcd. for C₁₂H₃NOS: C, 66.97; H, 4.18; N, 6.51; O, 7.44; S, 14.88. Found: C, 66.82; H, 4.16; N, 6.30; O, 7.63; S, 14.75.

N-(p-Tolyl)-5,6-dihydro-4-oxo-4H-thieno[3,4-c]pyrrole 2b.

Compound **2b** was obtained using the general procedure described above. Elution with ether/hexane (1/1) mixture and recrystallization from methylene chloride/hexane gave **2b** in 79% yield, mp 190-191°; ir: ν C=O 1688.0 cm⁻¹; pmr: δ 7.82 (d, H₃, 2.3 Hz), 7.16 (dt, H₁, 2.3 and 1.2 Hz), 7.21-7.63 (M, C_sH_s), 4.73 (d, CH₂, 1.2 Hz), 2.35 (s, CH₃).

Anal. Calcd. for C₁₈H₁₁NOS: C, 68.12; H, 4.80; N, 6.11; O, 6.98; S, 13.97. Found: C, 68.46; H, 4.78; N, 5.99; O, 7.08; S, 14.08.

N-Benzyl-5,6-dihydro-4-oxo-4H-thieno[3,4-c]pyrrole 2c.

Compound 2c was obtained using the general procedure described above. Elution with ether/hexane (2/1) mixture and recrystallization from methylene chloride/hexane gave 2c in 79% yield, mp 95-96°; ir: ν C=0 1678.2 cm⁻¹; pmr: δ 7.76 (d, H₃, 2.3 Hz), 7.03 (dt, H₁, 2.3 and 1.2 Hz), 7.20-7.40 (M, C₆H₂), 4.72 (s, CH₂(2)), 4.15 (d, CH₂(6), 1.2 Hz).

Anal. Calcd. for C₁₃H₁₁NOS: C, 68.12; H, 4.80; N, 6.11; O, 6.98; S, 13.97. Found: C, 68.32; H, 4.75; N, 6.29; O, 6.83; S, 13.70.

N-(1-Phenylethyl)-5,6-dihydro-4-oxo-4H-thieno[3,4-c]pyrrole 2d.

Compound **2d** was obtained using the general procedure described above. Elution with ether/hexane (2/1) mixture and recrystallization from methylene chloride/hexane gave **2d** in 73% yield, mp 104-105°; ir: ν C = O 1678.2 cm⁻¹; pmr: δ 7.74 (d, H₂, 2.3 Hz), 7.03 (ddd, H₁, 2.3, 1.2 and 1.2 Hz), 7.20-7.40 (M, C₆H₂), 5.74 (q, CH, 7.5 Hz), 1.66 (d, CH₃, 7.5 Hz), 4.22 (dd, CH(6), 15.5 and 1.2 Hz), 3.87 (dd, CH(6), 15.5 and 1.2 Hz).

Anal. Calcd. for C₁₄H₁₃NOS: C, 69.13; H, 5.35; N, 5.76; O, 6.58; S, 13.17. Found: C, 69.35; H, 5.27; N, 5.70; O, 6.88; S, 13.28.

N-Aryl-4-arylimino-5,6-dihydro-4H-thieno[3,4-c]pyrroles 3.

The same procedure as above (except 0.0055 mole of amine) led to compounds 3.

N-Phenyl-4-phenylimino-5,6-dihydro-4H-thieno[3,4-c]pyrrole 3a.

Compound 3a was obtained using the general procedure described above. Elution with ether/hexane (1/5) mixture and recrystallization from ether/hexane gave 3a in 57% yield, mp 148-149°; ir: ν C = N 1644.8 cm⁻¹;

pmr: δ 6.29 (d, H₃, 2.4 Hz), 7.00 (dt, H₁, 2.4 and 1.3 Hz), 7.00-7.96 (M, C_sH_s), 4.82 (d, CH₂, 1.3 Hz).

Anal. Calcd. for $C_{16}H_{14}N_2OS$: C, 74.48; H, 4.83; N, 9.66; S, 11.03. Found: C, 74.59; H, 4.91; N, 9.36; S, 11.27.

N-(p-Tolyl)-4-(p-tolyl)imino-5,6-dihydro-4H-thieno[3,4-c]pyrrole 3b.

Compound 3b was obtained using the general procedure described above. Elution with ether/hexane (1/5) mixture and recrystallization from ether/hexane gave 3b in 67% yield, mp 167-168°; ir: ν C=N 1638.1 cm⁻¹; pmr: δ 6.32 (d, H₃, 2.3 Hz), 7.00 (dt, H₁, 2.3 and 1.2 Hz), 6.80-7.80 (M, C_eH₄), 4.77 (d, CH₂, 1.2 Hz), 2.36 (s, CH₃), 2.33 (s, CH₃).

Anal. Calcd. for $C_{20}H_{18}N_2S$: C, 75.47; H, 5.66; N, 8.81; S, 10.06. Found: C, 75.07; H, 5.78; N, 8.55; S, 9.84.

N-Benzyl-4-benzylimino-5,6-dihydro-4H-thieno[3,4-c]pyrrole 3c.

Compound **3c** was obtained using the general procedure described above. Elution with ether/hexane (3/1) mixture and recrystallization from ether/hexane gave **3c** in 66% yield, mp 76-77°; ir: ν C = N 1644.8 cm⁻¹; pmr: δ 7.66 (d, H₃, 2.3 Hz), 6.98 (dt, H₁, 8.3 and 1.3 Hz), 7.20-7.50 (M, C₆H₃), 4.82 (s, CH₂), 5.00 (s, CH₂), 4.18 (d, CH₃(6), 1.3 Hz).

Anal. Calcd. for C₂₀H₁₈N₂S: C, 75.47; H, 5.66; N, 8.81; S, 10.06. Found: C, 75.42; H, 5.69; N, 8.69; S, 10.16.

N-(1-Phenylethyl)-4-(1-phenylethyl)imino-5,6-dihydro-4H-thieno[3,4-c]pyrrole 3d.

Compound **3d** was obtained using the general procedure described above. Elution with ether/hexane mixture gave **3d** as an oily compound in 40% yield; ir: ν C = N 1638.1 cm⁻¹; pmr: δ 7.59 (d, H₃, 2.3 Hz), 6.89 (ddd, H₁, 2.3, 1.2 and 1.2 Hz), 7.20-7.60 (M, C₆H₅), 5.90 (q, CH, 7.2 Hz), 5.27 (q, CH, 6.4 Hz), 1.60 (d, CH₃, 7.2 Hz), 1.54 (d, CH₃, 6.4 Hz), 4.15 (dd, CH(6), 14.7 and 1.2 Hz), 3.86 (dd, CH(6), 14.7 and 1.2 Hz).

Anal. Calcd. for C₂₂H₂₂N₂S: C, 76.30; H, 6.36; N, 8.09; S, 9.25. Found: C, 76.59; H, 6.38; N, 7.86; S, 9.11.

N-Aryl-5,6-dihydro-6-oxo-4H-thieno[2,3-c]pyrroles 5.

Compounds 5 are obtained from thiophene-2,3-dicarbaldehyde 4 using the procedure described for 2 without 2-mercaptoethanol.

N-Phenyl-5,6-dihydro-6-oxo-4H-thieno[2,3-c]pyrrole 5a.

Compound 5a was obtained using the general procedure described above. Elution with ether/hexane mixture and recrystallization from methylene chloride/hexane gave 5a in 27% yield, mp 161-162°; ir: ν C=0 1671.0 cm⁻¹; pmr: δ 7.75 (d, H₂, 4.7 Hz), 7.10 (d, H₃, 4.7 Hz), 7.20-7.50 (M, C_sH_s), 4.79 (s, CH₂).

Anal. Calcd. for $C_{12}H_9NOS$: C, 66.97; H, 4.18; N, 6.51; O, 7.44; S, 14.88. Found: C, 67.12; H, 4.34; N, 6.41; O, 7.65; S, 14.65.

N-(p-Tolyl)-5,6-dihydro-6-oxo-4H-thieno[2,3-c]pyrrole 5 \mathbf{b} .

Compound **5b** was obtained using the general procedure described above. Elution with ether/hexane (1/1) mixture and recrystallization from methylene chloride/hexane gave **5b** in 24% yield, mp 161-162°; ir: ν C=0 1675.0 cm⁻¹; pmr: δ 7.68 (d, H₂, 4.8 Hz), 7.09 (d, H₃, 4.8 Hz), 7.20 (d, C₄H₄, 7.1 Hz), 7.62 (d, C₄H₄, 7.1 Hz), 4.75 (s, CH₂), 2.29 (s, CH₃).

Anal. Calcd. for C₁₃H₁₁NOS: C, 68.12; H, 4.80; N, 6.11; O, 6.98; S, 13.97. Found: C, 68.05; H, 4.89; N, 5.98; O, 6.98; S, 14.19.

N-Benzyl-5,6-dihydro-6-oxo-4H-thieno[2,3-c]pyrrole 5c.

Compound 5c was obtained using the general procedure described above. Elution with ether/hexane (1/1) mixture and recrystallization from ether/hexane gave 5c in 32% yield, mp 102-103°; ir: ν C= O 1670.0 cm⁻¹; pmr: δ 7.64 (d, H₂, 4.7 Hz), 7.00 (d, H₃, 4.7 Hz), 7.20-7.40 (M, C₆H₈), 4.75 (s, CH₂), 4.18 (s, CH₂ (4)).

Anal. Calcd. for C₁₃H₁₂NOS: C, 68.12; H, 4.80; N, 6.11; O, 6.98; S, 13.97. Found: C, 68.29; H, 4.71; N, 6.11; O, 6.98; S, 14.38.

For the labelled compound **5d**, the same procedure was used starting from the appropriate deuterated thiophenedicarbaldehyde [10].

N-Benzyl-5,6-dihydro-6-oxo-4H-thieno[2,3-c]pyrrole (4-D) 5d.

Compound 5d was obtained using the general procedure described above. Elution with ether/hexane (1/1) mixture and recrystalllization from ether/hexane gave 5d in 21% yield, mp 101-102°; ir: ν C=0 1670.0 cm⁻¹; pmr: δ 7.61 (d, H₂, 4.7 Hz), 6.97 (d, H₃, 4.7 Hz), 7.20-7.40 (M, C₆H₃), 4.75 (s, CH₂), 4.15 (s, CH(4)).

Anal. Calcd. for $C_{18}H_{10}DNOS$: C, 67.50; H + D/2, 5.22; N, 6.05; S, 13.86. Found: C, 67.50; H + D/2, 4.82; N, 5.98; S, 13.75.

N(1-Phenylethyl)-5,6-dihydro-6-oxo-4H-thieno[2,3-c]pyrrole 5e.

Compound 5e was obtained using the general procedure described above. Elution with ether/hexane (1/1) mixture and recrystallization from diisopropylether/hexane gave 5e in 23% yield, mp 80-81°; ir: ν C=0 1671.5 cm⁻¹; pmr: δ 7.60 (d, H₂, 4.7 Hz), 6.96 (d, H₃, 4.7 Hz), 7.30-7.40 (M, C₆H₈), 5.71 (q, CH, 7.2 Hz), 1.68 (d, CH₃, 7.2 Hz).

Anal. Calcd. for C₁₄H₁₈NOS: C, 69.13; H, 5.35; N, 5.65; O, 6.58; S, 13.17. Found: C, 69.15; H, 5.48; N, 5.65; O, 6.87; S, 13.39.

Preparation of Compound 2b via 6.

A solution of p-toluidine (0.0025 mole, 0.30 g) in 50 ml of absolute ethanol was added dropwise at 0° to a mixture of 1 (0.0025 mole, 0.35 g) and 1,2-ethanedithiol (0.0025 mole, 0.20 ml) in 100 ml of absolute ethanol. The reaction mixture was stirred for 3 hours at room temperature, acidified with dilute hydrochloric acid, and extracted with ether. Recrystallization from benzene/hexane afforded 0.31 g (41%) of 6, mp 208°; pmr: δ 2.30 (s, CH₃), 2.65 (broad s, SCH₂CH₂S), 5.99 (s, CH₂), 6.89 (d, ortho-H, 7.3 Hz), 7.04 (s, thio-H), 7.14 (d, meta-H, 7.3 Hz).

Anal. Calcd. for C₁₅H₁₅NS₃: C, 59.01; H, 4.91; N, 4.59; S, 31.47. Found:

C, 59.14; H, 4.93; N, 4.53; S, 31.82.

This later compound were chromatographed on a silica gel column (eluting mixture: ether/hexane, 1/1) giving 2b in quantitative yield.

REFERENCES AND NOTES

- [1] G. Nannini, P. N. Giraldi, G. Molgora, G. Biasoli, F. Spinelli, W. Logemann, E. Dradi, G. Zanni, A. Buttinoni and R. Tommasini, *Arzneim.-Forsch.*, 23, 1090 (1973).
- [2] A. Buttinoni, A. Cuttica, J. Franceschini, V. Mandelli, G. Orsini, N. Passerini, C. Turba and R. Tommasini, ibid., 23, 1100 (1973).
- [3] T. Kametani, K. Kigasawa, M. Hiiragi, H. Ishimaru, S. Haga and K. Shirayama, *J. Heterocyclic Chem.*, 15, 369 (1978).
 - [4] M. J. Cook and E. J. Forbes, Tetrahedron, 24, 4501 (1968).
- [5a] M. Robba, R. C. Moreau and B. Roques, C. R. Acad. Sci., 259, 3568 (1964); [b] P. Pastour, P. Savalle and P. Eymery, C. R. Acad. Sci., 260, 6130 (1965).
 - [6] M. Farnier and P. Fournari, C. R. Acad. Sci., 277, 803 (1973).
- [7] T. Dominh, A. L. Johnson, J. E. Jones and P. P. Senise, Jr., J. Org. Chem., 42, 4217 (1977).
- [8] S. Nan'ya, T. Tange and E. Mackawa, J. Heterocyclic Chem., 22, 449 (1985).
- [9] R. Grigg, H. Q. Nimal Gunaratne and V. Sridharan, J. Chem. Soc., Chem. Commun., 1183 (1985).
- [10] Syntheses of labelled thiophenedicarbaldehydes were achieved via the n-Buli procedure (R. Guilard, These d'Etat, 1971, Dijon) using (H₃C)₂N-CDO as the formylating agent.